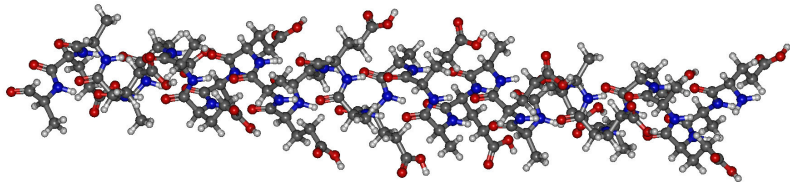


Large scale electronic structure calculations



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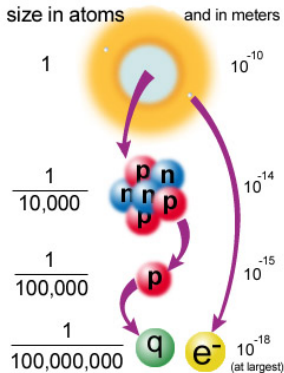
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What is quantum chemistry?

Quantum chemistry - understanding properties of matter at a subatomic scale.

Which scales we are talking about:

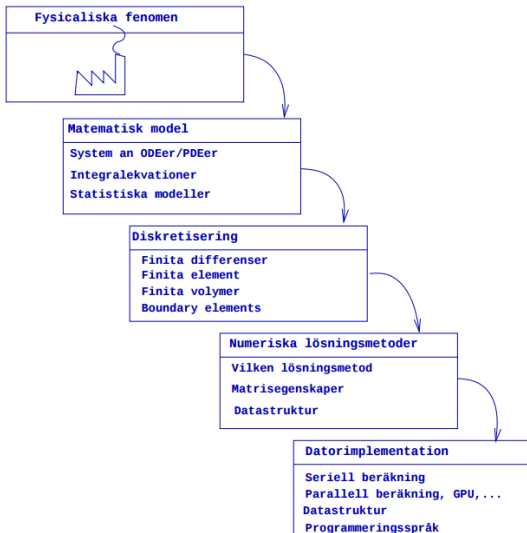
- electronic cloud of an isolated atom: $1 \text{ \AA} = 10^{-10} \text{ m}$
- size of the nucleus: 10^{-14} m (10000 times smaller!)
- vibration period of the molecular bond: $1 \text{ fs} = 10^{-15} \text{ s}$



(<http://www.particleadventure.org/>)

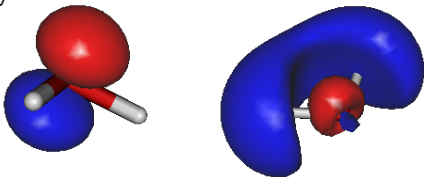
Quantum chemistry calculations:
modelling very small size systems during very short time

Modelling steps



Applications

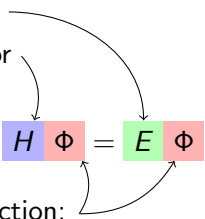
- modelling complex biological systems (DNA studies)
- drug design (preselecting molecules as potential good candidates)
- materials industry (selecting materials with the right properties)
- chemical industry
- food industry
- ...



Mathematical model

The **Schrödinger equation** rules the interactions between nuclei and electrons.

Non-relativistic time independent form (eigenvalue problem):

- Ground-state energy
 - Hamiltonian operator
- 
- The diagram shows the equation $H\Phi = E\Phi$ where H is in a blue box, Φ is in a red box, E is in a green box, and Φ is in a red box. Arrows indicate the following connections: from 'Ground-state energy' to E ; from 'Hamiltonian operator' to H ; from 'Many-body wavefunction:' to the Φ on the right; and a curved arrow from 'Ground-state energy' to the Φ on the left.
- Many-body wavefunction:
 $\Phi = \Phi(\text{coord. of nuclei, coord. of electrons})$

Mathematical model

Born-Oppenheimer approximation:

Assume that nuclei are fixed and do not move. Compute electronic wavefunction.

Then the total ground state energy is

$$E = E_{\text{electrons}} + E_{\text{nuclei}}$$

Electronic structure is the state of motion of electrons in an electrostatic field created by stationary nuclei.

Electronic structure calculations: numerically determination of electronic structure and predicting properties of matter from it.

Mathematical model

$E_{\text{electrons}} - ?$

Electronic Schrödinger equation:

$$H_{el}\Phi = E\Phi, \quad \Phi = \Phi(\text{coord. of electrons})$$

Mathematical model

$E_{\text{electrons}} - ?$

Electronic Schrödinger equation:

$$H_{el}\Phi = E\Phi, \quad \Phi = \Phi(\text{coord. of electrons})$$

Another approximation:

assume the form of the wavefunction – Φ expressed using single-particle wavefunctions ψ_i

Numerical discretization

Discretization of the single-particle wavefunction ψ_i

- Finite difference
- Finite elements; **localized basis functions**

Chosen basis set $\{\chi_j\}_{j=1}^m$

$$\psi_i = \sum_{j=1}^m c_{ji} \chi_j, \quad i = 1, \dots, N$$

Solution method

Our non-linear eigenvalue problem:

$$\hat{H}(C)C = EC$$

- (1) fix C , compute \hat{H}
- (2) solve $\hat{H}C = EC$ to get new C
- (3) if not converged, return to step (1)

Solution method

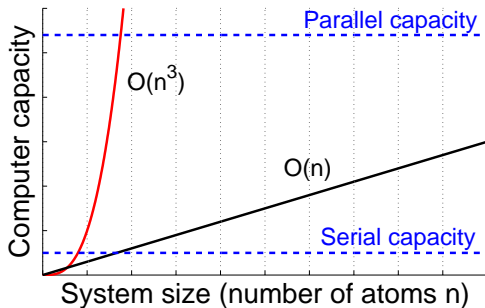
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Direct diagonalization is too expensive!
We want linear scaling!

Implementation



- quantum chemistry program “Ergo”
- Chunks and Tasks programming model

Plans and goals



- Parallelize Ergo using Chunks and Tasks
- Controlling errors due to approximations, only one threshold value!

Thank you for your attention!

